

(12) INTERNATIONAL APPLICATION PUBLISHED UNDER THE PATENT COOPERATION TREATY (PCT)

(19) World Intellectual Property Organization  
International Bureau



(43) International Publication Date  
28 November 2002 (28.11.2002)

PCT

(10) International Publication Number  
**WO 02/094215 A2**

(51) International Patent Classification<sup>7</sup>: **A61K 9/00**

(21) International Application Number: PCT/US01/43111

(22) International Filing Date:  
16 November 2001 (16.11.2001)

(25) Filing Language: English

(26) Publication Language: English

(30) Priority Data:  
09/715,117 20 November 2000 (20.11.2000) US

(71) Applicant (for all designated States except US): **ELAN PHARMA INTERNATIONAL LTD.** [IE/IE]; Wil House, Shannon Business Park, Shannon, County Clare (IE).

(72) Inventors; and

(75) Inventors/Applicants (for US only): **BOSCH, Henry, William** [US/US]; 237 Rodney Circle, Bryn Mawr, PA 19010 (US). **RYDE, Niels, P.** [SE/US]; 54 Lloyd Ave., Malvern, PA 19355 (US).

(74) Agents: **SIMKIN, Michele, M.** et al.; Foley & Lardner, Washington Harbour, 3000 K Street, NW, Suite 500, Washington, DC 20007-5109 (US).

(81) Designated States (national): AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, BZ, CA, CH, CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NO, NZ, OM, PH, PL, PT, RO, RU, SD, SE, SG, SI, SK, SL, TJ, TM, TR, TT, TZ, UA, UG, US, UZ, VN, YU, ZA, ZM, ZW.

(84) Designated States (regional): ARIPO patent (GH, GM, KE, LS, MW, MZ, SD, SL, SZ, TZ, UG, ZM, ZW), Eurasian patent (AM, AZ, BY, KG, KZ, MD, RU, TJ, TM), European patent (AT, BE, CH, CY, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE, TR), OAPI patent (BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG).

**Published:**

— without international search report and to be republished upon receipt of that report

For two-letter codes and other abbreviations, refer to the "Guidance Notes on Codes and Abbreviations" appearing at the beginning of each regular issue of the PCT Gazette.



**WO 02/094215 A2**

(54) Title: NANOPARTICULATE COMPOSITIONS COMPRISING COPOLYMERS AS SURFACE STABILIZERS

(57) **Abstract:** The present invention is directed to nanoparticulate compositions comprising a poorly soluble drug and at least one copolymer of vinyl pyrrolidone and vinyl acetate as a surface stabilizer adsorbed to the surface of the drug. Also encompassed by the invention are pharmaceutical compositions comprising a nanoparticulate composition of the invention, methods of making and using such nanoparticulate and pharmaceutical compositions.

## NANOPARTICULATE COMPOSITIONS COMPRISING COPOLYMERS AS SURFACE STABILIZERS

5

### FIELD OF THE INVENTION

The present invention is directed to nanoparticulate formulations of a drug having at least one copolymer of vinyl pyrrolidone and vinyl acetate adsorbed on the surface of the drug as a surface stabilizer, and methods of making and using such compositions.

10

### BACKGROUND OF THE INVENTION

Nanoparticulate compositions, first described in U.S. Patent No. 5,145,684 ("the '684 patent"), are particles consisting of a poorly soluble therapeutic or diagnostic agent having adsorbed onto the surface thereof a non-crosslinked surface stabilizer. The 15 '684 patent describes the use of a variety of surface stabilizers for nanoparticulate compositions. The use of a copolymer of vinyl pyrrolidone and vinyl acetate as a surface stabilizer for nanoparticulate compositions, or any other component of such compositions, is not described by the '684 patent.

The '684 patent describes a method of screening drugs to identify useful 20 surface stabilizers that enable the production of a nanoparticulate composition. Not all surface stabilizers will function to produce a stable, non-agglomerated nanoparticulate composition for all drugs. Moreover, known surface stabilizers may be unable to produce a stable, non-agglomerated nanoparticulate composition for certain drugs. Thus, there is a need in the art to identify new surface stabilizers useful in making nanoparticulate 25 compositions. Additionally, such new surface stabilizers may have superior properties over prior known surface stabilizers.

There is a need in the art for new surface stabilizers for nanoparticulate compositions of poorly soluble drugs. In addition, there is a need in the art for surface stabilizers useful in preparing nanoparticulate compositions of drugs, in which prior 30 known surface stabilizers are ineffective. The present invention satisfies these needs.

### SUMMARY OF THE INVENTION

The present invention is directed to nanoparticulate compositions comprising a poorly soluble drug and at least one copolymer of vinyl pyrrolidone and vinyl acetate as a surface stabilizer adsorbed to the surface of the drug.

Another aspect of the invention is directed to pharmaceutical compositions comprising a nanoparticulate composition of the invention. The pharmaceutical composition preferably comprises a poorly soluble drug, at least one copolymer of vinyl pyrrolidone and vinyl acetate as a surface stabilizer adsorbed to the surface of the drug, and a pharmaceutically acceptable carrier, as well as any desired excipients.

This invention further discloses a method of making a nanoparticulate composition having at least one copolymer of vinyl pyrrolidone and vinyl acetate as a surface stabilizer adsorbed on the surface of the drug. Such a method comprises contacting a poorly soluble nanoparticulate drug with at least one copolymer of vinyl pyrrolidone and vinyl acetate as a surface stabilizer for a time and under conditions sufficient to provide a nanoparticle/copolymer composition. The copolymer surface stabilizers can be contacted with the drug either before, during, or after size reduction of the drug.

The present invention is further directed to a method of treatment comprising administering to a mammal in need a therapeutically effective amount of a nanoparticulate drug/copolymer composition according to the invention.

Both the foregoing general description and the following detailed description are exemplary and explanatory and are intended to provide further explanation of the invention as claimed. Other objects, advantages, and novel features will be readily apparent to those skilled in the art from the following detailed description of the invention.

## **DETAILED DESCRIPTION OF THE INVENTION**

The present invention is directed to a composition comprising nanoparticulate drug having at least one copolymer of vinyl pyrrolidone and vinyl acetate 5 as a surface stabilizer adsorbed on the surface thereof, and methods of making and using such nanoparticulate compositions.

### **A. Compositions**

The compositions of the invention comprise nanoparticulate drug and at 10 least one copolymer of vinyl pyrrolidone and vinyl acetate as a surface stabilizer adsorbed to the surface of the drug. Surface stabilizers useful herein physically adhere to the surface of the nanoparticulate drug, but do not chemically react with the drug or itself. Individually adsorbed molecules of the surface stabilizer are essentially free of intermolecular cross-linkages.

The present invention also includes nanoparticulate compositions having at 15 least one copolymer of vinyl pyrrolidone and vinyl acetate as a surface stabilizer adsorbed on the surface thereof, formulated into compositions together with one or more non-toxic physiologically acceptable carriers, adjuvants, or vehicles, collectively referred to as carriers. The compositions can be formulated for parenteral injection, oral administration 20 in solid or liquid form, rectal or topical administration, and the like.

#### **1. Drug Particles**

The nanoparticles of the invention comprise a therapeutic or diagnostic agent, collectively referred to as a "drug." A therapeutic agent can be a pharmaceutical 25 agent, including biologics such as proteins, peptides, and nucleotides, or a diagnostic agent, such as a contrast agent, including x-ray contrast agents. The drug exists either as a discrete, crystalline phase, or as an amorphous phase. The crystalline phase differs from a non-crystalline or amorphous phase which results from precipitation techniques, such as those described in EP Patent No. 275,796.

The invention can be practiced with a wide variety of drugs. The drug is 30 preferably present in an essentially pure form, is poorly soluble, and is dispersible in at

least one liquid medium. By "poorly soluble" it is meant that the drug has a solubility in the liquid dispersion medium of less than about 10 mg/mL, and preferably of less than about 1 mg/mL.

The drug can be selected from a variety of known classes of drugs,  
5 including, for example, proteins, peptides, nucleotides, anti-obesity drugs, nutriceuticals, corticosteroids, elastase inhibitors, analgesics, anti-fungals, oncology therapies, anti-emetics, analgesics, cardiovascular agents, anti-inflammatory agents, anthelmintics, anti-arrhythmic agents, antibiotics (including penicillins), anticoagulants, antidepressants, antidiabetic agents, antiepileptics, antihistamines, antihypertensive agents, antimuscarinic agents, antimycobacterial agents, antineoplastic agents, immunosuppressants, antithyroid agents, antiviral agents, anxiolytic sedatives (hypnotics and neuroleptics), astringents, beta-adrenoceptor blocking agents, blood products and substitutes, cardiac inotropic agents, contrast media, corticosteroids, cough suppressants (expectorants and mucolytics), diagnostic agents, diagnostic imaging agents, diuretics, dopaminergics (antiparkinsonian agents), haemostatics, immunochemical agents, lipid regulating agents, muscle relaxants, parasympathomimetics, parathyroid calcitonin and biphosphonates, prostaglandins, radio-pharmaceuticals, sex hormones (including steroids), anti-allergic agents, stimulants and anoretics, sympathomimetics, thyroid agents, vasodilators and xanthines.

The drugs are commercially available and/or can be prepared by  
20 techniques known in the art.

## 2. Copolymer Surface Stabilizers

Commercially available copolymers of vinyl pyrrolidone and vinyl acetate are Plasdone® S630 (ISP) and Kollidon® VA 64 (BASF), which contain vinyl pyrrolidone and vinyl acetate in a 60:40 ratio. Other copolymers of vinyl pyrrolidone and vinyl acetate can also be used in the invention. Preferably, the copolymer contains at least 40% vinyl pyrrolidone, although smaller amounts of vinyl pyrrolidone can also be utilized. Other useful copolymers contain vinyl pyrrolidone and vinyl acetate in ratios of, for example, 90:10, 80:20, 70:30, and 50:50. The amount of vinyl pyrrolidone can range

from about 40% up to about 99.9%, and the amount of vinyl acetate can range from about 0.1% up to about 60%.

Two or more surface stabilizers can be used in combination.

### 5        3. Auxiliary Surface Stabilizers

The compositions of the invention can also include one or more auxiliary surface stabilizers in addition to the at least one copolymer of vinyl pyrrolidone and vinyl acetate. Suitable auxiliary surface stabilizers can preferably be selected from known organic and inorganic pharmaceutical excipients. Such excipients include various polymers, low molecular weight oligomers, natural products, and surfactants. Preferred surface stabilizers include nonionic and ionic surfactants. Two or more surface auxiliary stabilizers can be used in combination.

Representative examples of auxiliary surface stabilizers include cetyl pyridinium chloride, gelatin, casein, lecithin (phosphatides), dextran, glycerol, gum acacia, cholesterol, tragacanth, stearic acid, benzalkonium chloride, calcium stearate, glycerol monostearate, cetostearyl alcohol, cetomacrogol emulsifying wax, sorbitan esters, polyoxyethylene alkyl ethers (*e.g.*, macrogol ethers such as cetomacrogol 1000), polyoxyethylene castor oil derivatives, polyoxyethylene sorbitan fatty acid esters (*e.g.*, the commercially available Tweens® such as *e.g.*, Tween 20® and Tween 80® (ICI Specialty Chemicals)); polyethylene glycols (*e.g.*, Carbowaxes 3350® and 1450®, and Carbopol 934® (Union Carbide)), dodecyl trimethyl ammonium bromide, polyoxyethylene stearates, colloidal silicon dioxide, phosphates, sodium dodecylsulfate, carboxymethylcellulose calcium, hydroxypropyl celluloses (*e.g.*, HPC, HPC-SL, and HPC-L), hydroxypropyl methylcellulose (HPMC), carboxymethylcellulose sodium, methylcellulose, hydroxyethylcellulose, hydroxypropylmethyl-cellulose phthalate, noncrystalline cellulose, magnesium aluminum silicate, triethanolamine, polyvinyl alcohol (PVA), polyvinylpyrrolidone (PVP), 4-(1,1,3,3-tetramethylbutyl)-phenol polymer with ethylene oxide and formaldehyde (also known as tyloxapol, superione, and triton), poloxamers (*e.g.*, Pluronics F68® and F108®, which are block copolymers of ethylene oxide and propylene oxide); poloxamines (*e.g.*, Tetronic 908®, also known as Poloxamine

908<sup>®</sup>, which is a tetrafunctional block copolymer derived from sequential addition of propylene oxide and ethylene oxide to ethylenediamine (BASF Wyandotte Corporation, Parsippany, N.J.)); a charged phospholipid such as dimyristoyl phophatidyl glycerol, dioctylsulfosuccinate (DOSS); Tetronic 1508<sup>®</sup>(T-1508) (BASF Wyandotte Corporation ), 5 dialkylesters of sodium sulfosuccinic acid (e.g., Aerosol OT<sup>®</sup>, which is a dioctyl ester of sodium sulfosuccinic acid (Cytec Industries, West Paterson, NJ)); Duponol P<sup>®</sup>, which is a sodium lauryl sulfate (DuPont); Triton X-200<sup>®</sup>, which is an alkyl aryl polyether sulfonate (Union Carbide); Crodestas F-110<sup>®</sup>, which is a mixture of sucrose stearate and sucrose distearate (Croda Inc.); p-isonylphenoxy poly-(glycidol), also known as Olin-LOG<sup>®</sup> or 10 Surfactant 10-G<sup>®</sup> (Olin Chemicals, Stamford, CT); Crodestas SL-40<sup>®</sup> (Croda, Inc.); decanoyl-N-methylglucamide; n-decyl  $\beta$ -D-glucopyranoside; n-decyl  $\beta$ -D-maltopyranoside; n-dodecyl  $\beta$ -D-glucopyranoside; n-dodecyl  $\beta$ -D-maltoside; heptanoyl 15 N-methylglucamide; n-heptyl- $\beta$ -D-glucopyranoside; n-heptyl  $\beta$ -D-thioglucoside; n-hexyl  $\beta$ -D-glucopyranoside; nonanoyl-N-methylglucamide; n-noyl  $\beta$ -D-glucopyranoside; octanoyl-N-methylglucamide; n-octyl- $\beta$ -D-glucopyranoside; octyl  $\beta$ -D-thioglucopyranoside; and the like.

Most of these surface stabilizers are known pharmaceutical excipients and are described in detail in the *Handbook of Pharmaceutical Excipients*, published jointly by the American Pharmaceutical Association and The Pharmaceutical Society of Great 20 Britain (The Pharmaceutical Press, 2000), specifically incorporated by reference. The surface stabilizers are commercially available and/or can be prepared by techniques known in the art.

#### 4. Nanoparticulate Drug/Copolymer Particle Size

25 Preferably, the compositions of the invention contain nanoparticles which have an effective average particle size of less than about 2000 nm (*i.e.*, 2 microns), more preferably less than about 1500 nm, less than about 1000 nm, less than about 800 nm, less than about 600 nm, less than about 400 nm, less than about 300 nm, less than about 250 nm, less than about 100 nm, or less than about 50 nm, as measured by light-scattering

methods, microscopy, or other appropriate methods. By "an effective average particle size of less than about 2000 nm" it is meant that at least 50% of the drug particles have a particle size of less than about 2000 nm when measured by light scattering techniques. Preferably, at least 70% of the drug particles have a particle size of less than about 2000 nm, more preferably at least 90% of the drug particles have a particle size of less than about 2000 nm, and even more preferably at least about 95% of the particles have a particle size of less than about 2000 nm.

### 5. Concentration of Nanoparticulate Drug and Stabilizer

The relative amount of drug and one or more surface stabilizers can vary widely. The optimal amount of the surface stabilizers can depend, for example, upon the particular active agent selected, the hydrophilic lipophilic balance (HLB), melting point, and water solubility of the copolymer, and the surface tension of water solutions of the stabilizer, *etc.*

The concentration of the one or more surface stabilizers can vary from about 0.01 to about 90%, from about 1 to about 75%, from about 10 to about 60%, and from about 10 to about 55% by weight based on the total combined weight of the drug substance and surface stabilizer, not including other excipients.

The concentration of the drug can vary from about 99.9% to about 0.1%, from about 80% to about 5.0%, or from about 50% to about 10%, by weight based on the total combined weight of the drug substance and surface stabilizer, not including other excipients.

### B. Methods of Making Nanoparticulate Formulations

The nanoparticulate drug compositions can be made using, for example, milling or precipitation techniques. Exemplary methods of making nanoparticulate compositions are described in the '684 patent. Methods of making nanoparticulate compositions are also described in U.S. Patent No. 5,518,187, for "Method of Grinding Pharmaceutical Substances;" U.S. Patent No. 5,718,388, for "Continuous Method of Grinding Pharmaceutical Substances;" U.S. Patent No. 5,862,999, for "Method of Grinding Pharmaceutical Substances;" U.S. Patent No. 5,665,331, for "Co-

Microprecipitation of Nanoparticulate Pharmaceutical Agents with Crystal Growth Modifiers;" U.S. Patent No. 5,662,883, for "Co-Microprecipitation of Nanoparticulate Pharmaceutical Agents with Crystal Growth Modifiers;" U.S. Patent No. 5,560,932, for "Microprecipitation of Nanoparticulate Pharmaceutical Agents;" U.S. Patent No. 5,543,133, for "Process of Preparing X-Ray Contrast Compositions Containing Nanoparticles;" U.S. Patent No. 5,534,270, for "Method of Preparing Stable Drug Nanoparticles;" U.S. Patent No. 5,510,118, for "Process of Preparing Therapeutic Compositions Containing Nanoparticles;" and U.S. Patent No. 5,470,583, for "Method of Preparing Nanoparticle Compositions Containing Charged Phospholipids to Reduce Aggregation," all of which are specifically incorporated by reference..

### **1. Milling to obtain Nanoparticulate Drug Dispersions**

Milling of aqueous drug to obtain a nanoparticulate dispersion comprises dispersing drug particles in a liquid dispersion medium in which the drug is poorly soluble, followed by applying mechanical means in the presence of grinding media to reduce the particle size of the drug to the desired effective average particle size. The dispersion medium can be, for example, water, safflower oil, ethanol, t-butanol, glycerin, polyethylene glycol (PEG), hexane, or glycol.

The particles can be reduced in size in the presence of at least one copolymer of vinyl pyrrolidone and vinyl acetate surface stabilizer. Alternatively, the particles can be contacted with one or more surface stabilizers after attrition. Other compounds, such as a diluent, can be added to the drug/surface stabilizer composition during the size reduction process. Dispersions can be manufactured continuously or in a batch mode. The resultant nanoparticulate drug dispersion can be utilized in solid or liquid dosage formulations.

### **2. Precipitation to Obtain Nanoparticulate Drug Compositions**

Another method of forming the desired nanoparticulate composition is by microprecipitation. This is a method of preparing stable dispersions of poorly soluble

drugs in the presence of one or more surface stabilizers and one or more colloid stability enhancing surface active agents free of any trace toxic solvents or solubilized heavy metal impurities. Such a method comprises, for example: (1) dissolving the poorly soluble drug in a suitable solvent; (2) adding the formulation from step (1) to a solution comprising at least one copolymer surface stabilizer to form a clear solution; and (3) precipitating the formulation from step (2) using an appropriate non-solvent. The method can be followed by removal of any formed salt, if present, by dialysis or diafiltration and concentration of the dispersion by conventional means. The resultant nanoparticulate drug dispersion can be utilized in solid or liquid dosage formulations.

10

### C. Methods of Using Nanoparticulate Drug Formulations Comprising One or More Surface Stabilizers

The nanoparticulate compositions of the present invention can be administered to humans and animals via any conventional means, including but not limited to orally, rectally, parenterally (intravenous, intramuscular, or subcutaneous), intracisternally, pulmonary, intravaginally, intraperitoneally, locally (powders, ointments or drops), or as a buccal or nasal spray.

Compositions suitable for parenteral injection may comprise physiologically acceptable sterile aqueous or nonaqueous solutions, dispersions, suspensions or emulsions and sterile powders for reconstitution into sterile injectable solutions or dispersions. Examples of suitable aqueous and nonaqueous carriers, diluents, solvents, or vehicles including water, ethanol, polyols (propyleneglycol, polyethylene-glycol, glycerol, and the like), suitable mixtures thereof, vegetable oils (such as olive oil) and injectable organic esters such as ethyl oleate. Proper fluidity can be maintained, for example, by the use of a coating such as lecithin, by the maintenance of the required particle size in the case of dispersions, and by the use of surfactants.

The nanoparticulate compositions may also contain adjuvants such as preserving, wetting, emulsifying, and dispensing agents. Prevention of the growth of microorganisms can be ensured by various antibacterial and antifungal agents, such as parabens, chlorobutanol, phenol, sorbic acid, and the like. It may also be desirable to

include isotonic agents, such as sugars, sodium chloride, and the like. Prolonged absorption of the injectable pharmaceutical form can be brought about by the use of agents delaying absorption, such as aluminum monostearate and gelatin.

Solid dosage forms for oral administration include capsules, tablets, pills, 5 powders, and granules. In such solid dosage forms, the active compound is admixed with at least one of the following: (a) one or more inert excipients (or carrier), such as sodium citrate or dicalcium phosphate; (b) fillers or extenders, such as starches, lactose, sucrose, glucose, mannitol, and silicic acid; (c) binders, such as carboxymethylcellulose, alignates, gelatin, polyvinylpyrrolidone, sucrose and acacia; (d) humectants, such as glycerol; 10 (e) disintegrating agents, such as agar-agar, calcium carbonate, potato or tapioca starch, alginic acid, certain complex silicates, and sodium carbonate; (f) solution retarders, such as paraffin; (g) absorption accelerators, such as quaternary ammonium compounds; (h) wetting agents, such as cetyl alcohol and glycerol monostearate; (i) adsorbents, such as kaolin and bentonite; and (j) lubricants, such as talc, calcium stearate, magnesium 15 stearate, solid polyethylene glycols, sodium lauryl sulfate, or mixtures thereof. For capsules, tablets, and pills, the dosage forms may also comprise buffering agents.

Liquid dosage forms for oral administration include pharmaceutically acceptable emulsions, solutions, suspensions, syrups, and elixirs. In addition to the active compounds, the liquid dosage forms may comprise inert diluents commonly used in the 20 art, such as water or other solvents, solubilizing agents, and emulsifiers. Exemplary emulsifiers are ethyl alcohol, isopropyl alcohol, ethyl carbonate, ethyl acetate, benzyl alcohol, benzyl benzoate, propyleneglycol, 1,3-butyleneglycol, dimethylformamide, oils, such as cottonseed oil, groundnut oil, corn germ oil, olive oil, castor oil, and sesame oil, glycerol, tetrahydrofurfuryl alcohol, polyethyleneglycols, fatty acid esters of sorbitan, or 25 mixtures of these substances, and the like.

Besides such inert diluents, the composition can also include adjuvants, such as wetting agents, emulsifying and suspending agents, sweetening, flavoring, and perfuming agents.

Actual dosage levels of active ingredients in the nanoparticulate 30 compositions of the invention may be varied to obtain an amount of active ingredient that

is effective to obtain a desired therapeutic response for a particular composition and method of administration. The selected dosage level therefore depends upon the desired therapeutic effect, the route of administration, the potency of the administered drug, the desired duration of treatment, and other factors.

5           The total daily dose of the compounds of this invention administered to a host in single or divided dose may be in amounts of, for example, from about 1 nanomol to about 50 micromoles per kilogram of body weight. Dosage unit compositions may contain such amounts of such submultiples thereof as may be used to make up the daily dose. It will be understood, however, that the specific dose level for any particular patient  
10          will depend upon a variety of factors including the body weight, general health, sex, diet, time and route of administration, potency of the administered drug, rates of absorption and excretion, combination with other drugs and the severity of the particular disease being treated.

15

\* \* \* \* \*

The following examples are given to illustrate the present invention. It should be understood, however, that the invention is not to be limited to the specific conditions or details described in these examples. Throughout the specification, any and all references to a publicly available document, including a U.S. patent, are specifically  
20          incorporated by reference.

### Example 1

The purpose of this example was to prepare a nanoparticulate dispersion of a naproxen composition comprising a copolymer of vinyl pyrrolidone and vinyl acetate.

25          Naproxen is a non-steroidal anti-inflammatory drug (NSAID) used in the treatment of rheumatoid conditions.

5% (w/w) of naproxen and 1% Plasdone® S-630 (60% vinyl pyrrolidone, 40% vinyl acetate) (ISP Technologies, Inc.) was milled using a Dyno® Mill (Type: KDL; Mfg.: Willy A Bachofen AG, Basel Switzerland) equipped with a 150 cc batch chamber  
30          using a 500 µm milling media (PolyMill® 500; Dow Chemical) for 2 hrs at ca. 10°C

Following milling, the nanoparticulate naproxen dispersion had a mean particle size of 96 nm, with 90% of the particles having a size of less than 141 nm.

### Example 2

5       The purpose of this example was to prepare a nanoparticulate dispersion of a nifedipine composition comprising a copolymer of vinyl pyrrolidone and vinyl acetate. Nifedipine is a poorly water-soluble calcium channel blocking agent. The drug affects the movement of calcium into heart and blood vessel cells, and causes a relaxing effect of the muscles to allow an increased amount of blood flow into the heart. Nifedipine is useful in  
10      treating angina pectoris (chest pain), and to help reduce blood pressure (antihypertensive).

An aqueous solution of 1% Plasdone® S-630 (60% vinyl pyrrolidone and 40% vinyl acetate) (ISP Technologies, Inc.) and 0.05% sodium lauryl sulfate (SLS) (Spectrum) was prepared by dissolving 0.85 g of polymer and 4.59 g of a 1% SLS solution in 75.66 g of deionized water. The stabilizer solution was mixed with 4.25 g of  
15      nifedipine (5% w/w) and charged into the chamber of a DYNO®-Mill Type KDL media mill (Willy Bachofen AG, Basel, Switzerland) along with 500 micron polymeric media (PolyMill® 500; Dow Chemical). The mill was operated for 2 hours and yielded a stable colloidal dispersion of drug substance having a mean particle size of 132 nm, with 90% of the particles having a size of less than 193 nm.

20

### Example 3

The purpose of this example was to prepare a nanoparticulate dispersion of a ketoprofen composition comprising a copolymer of vinyl pyrrolidone and vinyl acetate. Ketoprofen is a nonsteroidal anti-inflammatory drug (NSAID) effective in treating fever,  
25      pain, and inflammation in the body.

An aqueous solution of 1% Plasdone® S-630 (60% vinyl pyrrolidone and 40% vinyl acetate) (ISP Technologies, Inc.) and 0.05% sodium lauryl sulfate (SLS) (Spectrum) was prepared by dissolving 0.85 g of polymer and 4.26 g of a 1% SLS solution in 75.71 g of deionized water. The stabilizer solution was mixed with 4.25 g of  
30      ketoprofen (Wyckoff; 5% w/w) and charged into the chamber of a DYNO®-Mill Type

KDL media mill (Willy Bachofen AG, Basel, Switzerland) along with 500 micron polymeric media (PolyMill® 500; Dow Chemical). The mill was operated for 1 hour and yielded a stable colloidal dispersion of drug substance having a mean particle size of 256 nm, with 90% of the particles having a size of less than 355 nm.

5

**Example 4**

The purpose of this example was to prepare a nanoparticulate dispersion of a triamcinolone acetonide composition comprising a copolymer of vinyl pyrrolidone and vinyl acetate. Triamcinolone acetonide is a corticosteroid used as an antiallergic agent.

An aqueous solution of 1% Plasdone® S-630 (60% vinyl pyrrolidone and 40% vinyl acetate) (ISP Technologies, Inc.) and 0.05% sodium lauryl sulfate (SLS) (Spectrum) was prepared by dissolving 0.85 g of polymer and 4.30 g of a 1% SLS solution in 76.10 g of deionized water. The stabilizer solution was mixed with 4.26 g of triamcinolone acetonide (5% w/w) and charged into the chamber of a DYNO®-Mill Type KDL media mill (Willy Bachofen AG, Basel, Switzerland) along with 500 micron polymeric media (PolyMill® 500; Dow Chemical). The mill was operated for 2 hours and yielded a colloidal dispersion of drug substance having a mean particle size of 121 nm, with 90% of the particles having a size of less than 194 nm.

20      **Example 5**

The purpose of this example was to prepare a nanoparticulate dispersion of a nanoparticulate diagnostic imaging agent, benzoic acid, 3,5-bis(acetylamino) 2,4,6-triodo, 4-(ethyl-3-ethoxy-2-butenoate) ester (WIN 68209) composition comprising a copolymer of vinyl pyrrolidone and vinyl acetate.

An aqueous solution of 1% of a copolymer of 80% vinyl pyrrolidone and 20% vinyl acetate (Polysciences Inc., Warrington, PA) was prepared by dissolving 0.85 g of polymer in 79.91 g of deionized water. The stabilizer solution was mixed with 4.26 g of WIN 68209 (5% drug) and charged into the chamber of a DYNO®-Mill Type KDL media mill (Willy Bachofen AG, Basel, Switzerland) along with 500 micron polymeric media (PolyMill 500; Dow Chemical). The mill was operated for 1 hour and yielded a

stable colloidal dispersion of WIN 68209 having a mean particle size of 242 nm, with 90% of the particles having a size of less than 347 nm.

#### Example 6

5           The purpose of this example was to prepare a nanoparticulate dispersion of a nanoparticulate diagnostic imaging agent, WIN 68209, composition comprising a copolymer of vinyl pyrrolidone and vinyl acetate.

An aqueous solution of 1% of a copolymer of 80% vinyl pyrrolidone and 20% vinyl acetate (Polysciences Inc., Warrington, PA) and 1% SLS was prepared by 10 dissolving 0.85 g of polymer and 4.25 g of a 1% SLS solution in 75.67 g of deionized water. The stabilizer solution was mixed with 4.26 g of WIN 68209 (5% drug) and charged into the chamber of a DYNO®-Mill Type KDL media mill (Willy Bachofen AG, Basel, Switzerland) along with 500 micron polymeric media (PolyMill 500; Dow Chemical). The mill was operated for 1 hour and yielded a stable colloidal dispersion of 15 WIN 68209 having a mean particle size of 188 nm, with 90% of the particles having a size of less than 308 nm.

#### Example 7

20           The purpose of this example was to prepare a nanoparticulate dispersion of a nanoparticulate diagnostic imaging agent, WIN 68209, composition comprising a copolymer of vinyl pyrrolidone and vinyl acetate.

An aqueous solution of 1% of a copolymer of 50% vinyl pyrrolidone and 50% vinyl acetate (Polysciences Inc., Warrington, PA) and 0.05% SLS (Spectrum) was prepared by dissolving 0.85 g of polymer and 0.043 g of SLS in 79.86 g of deionized 25 water. The stabilizer solution was mixed with 4.25 g of WIN 68209 (5% drug) and charged into the chamber of a DYNO®-Mill Type KDL media mill (Willy Bachofen AG, Basel, Switzerland) along with 500 micron polymeric media (PolyMill 500; Dow Chemical). The mill was operated for 2 hours and yielded a stable colloidal dispersion of 30 WIN 68209 having a mean particle size of 96 nm, with 90% of the particles having a size of less than 143 nm.

\* \* \* \*

It will be apparent to those skilled in the art that various modifications and  
variations can be made in the methods and compositions of the present invention without  
departing from the spirit or scope of the invention. Thus, it is intended that the present  
invention cover the modifications and variations of this invention provided they come  
within the scope of the appended claims and their equivalents.

We claim:

1. A nanoparticulate composition comprising a drug having at least one copolymer of vinyl pyrrolidone and vinyl acetate adsorbed on the surface thereof as a surface stabilizer in an amount sufficient to maintain an effective average particle size of less than about 2000 nm.  
5
2. The composition of claim 1, wherein the copolymer of vinyl pyrrolidone and vinyl acetate has from about 40% up to about 99.9% vinyl pyrrolidone and about 10 0.1% up to about 60% vinyl acetate.
3. The composition of claim 2, wherein the copolymer of vinyl pyrrolidone and vinyl acetate has from about 50% up to about 99.9% vinyl pyrrolidone and about 0.1% up to about 50% vinyl acetate.  
15
4. The composition of claim 2, wherein the copolymer of vinyl pyrrolidone and vinyl acetate has about 60% vinyl pyrrolidone and about 40% vinyl acetate.
5. The composition of claim 2, wherein the copolymer of vinyl pyrrolidone and vinyl acetate has about 80% vinyl pyrrolidone and about 20% vinyl acetate.  
20
6. The composition of claim 1, wherein the drug is present in an amount selected from the group consisting of from about 99.9% to about 0.1%, from about 80% to about 5.0%, and from about 50% to about 10%, by weight based on the total combined weight of the drug substance and surface stabilizer, not including other excipients.  
25
7. The composition of claim 1, wherein at least one copolymer of vinyl pyrrolidone and vinyl acetate is present in an amount selected from the group consisting of from about 0.1 to about 90%, from about 1 to about 75%, from about 10 to about 60%,

and from about 10 to about 55%, by weight based on the total combined weight of the drug substance and surface stabilizer, not including other excipients.

8. The composition of claim 1, wherein the drug is selected from the group  
5 consisting of a crystalline phase drug and an amorphous phase drug.

9. The composition of claim 1, further comprising at least one surface stabilizer which is not a copolymer of vinyl pyrrolidone and vinyl acetate.

10 10. The composition of claim 1, comprising two or more copolymers of vinyl pyrrolidone and vinyl acetate.

11. The composition of claim 1, wherein the effective average particle size of the nanoparticulate composition is less than about 1500 nm.

15 12. The composition of claim 11, wherein the effective average particle size of the nanoparticulate composition is less than about 1000 nm.

20 13. The composition of claim 12, wherein the effective average particle size of the nanoparticulate composition is less than about 800 nm.

14. The composition of claim 13, wherein the effective average particle size of the nanoparticulate composition is less than about 600 nm.

25 15. The composition of claim 14, wherein the effective average particle size of the nanoparticulate composition is less than about 400 nm.

16. The composition of claim 15, wherein the effective average particle size of the nanoparticulate composition is less than about 300 nm.

17. The composition of claim 16, wherein the effective average particle size of the nanoparticulate composition is less than about 200 nm.

18. The composition of claim 17, wherein the effective average particle size of  
5 the nanoparticulate composition is less than about 100 nm.

19. The composition of claim 18, wherein the effective average particle size of the nanoparticulate composition is less than about 50 nm.

10 20. A method of making a nanoparticulate composition comprising an organic drug having at least one copolymer of vinyl pyrrolidone and vinyl acetate as a surface stabilizer adsorbed on the surface thereof in an amount sufficient to maintain an effective average particle size of less than about 2000 nm,

15 said method comprising contacting said drug with at least one copolymer of vinyl pyrrolidone and vinyl acetate for a time and under conditions sufficient to provide a nanoparticle/copolymer composition having an effective average particle size of less than about 2000 nm.

20 21. The method of claim 20, wherein the copolymer of vinyl pyrrolidone and vinyl acetate has from about 40% up to about 99.9% vinyl pyrrolidone and about 0.1% up to about 60% vinyl acetate.

25 22. The method of claim 20, wherein the drug is present in an amount selected from the group consisting of from about 99.9% to about 0.1%, from about 80% to about 5.0%, and from about 50% to about 10%, by weight based on the total combined weight of the drug substance and surface stabilizer, not including other excipients.

30 23. The method of claim 20, wherein the at least one copolymer of vinyl pyrrolidone and vinyl acetate is present in an amount selected from the group consisting of from about 0.1 to about 90%, from about 1 to about 75%, from about 10 to about 60%,

and from about 10 to about 55%, by weight based on the total combined weight of the drug substance and surface stabilizer, not including other excipients.

24. The method of claim 20, wherein the drug is selected from the group  
5 consisting of a crystalline phase drug and an amorphous phase drug.

25. The method of claim 20, further comprising at least one surface stabilizer which is not a copolymer of vinyl pyrrolidone and vinyl acetate.

10 26. The method of claim 20, comprising two or more copolymers of vinyl pyrrolidone and vinyl acetate.

27. The method of claim 20, wherein the effective average particle size of the nanoparticulate composition is selected from the group consisting of less than about 1500 nm, less than about 1000 nm, less than about 800 nm, less than about 600 nm, less than about 400 nm, less than about 300 nm, less than about 200 nm, less than about 100 nm, and less than about 50 nm.

28. A method of making a nanoparticulate composition, wherein the  
20 nanoparticulate composition comprises an organic drug having at least one copolymer of vinyl pyrrolidone and vinyl acetate as a surface stabilizer adsorbed on the surface thereof in an amount sufficient to maintain an effective average particle size of less than about 2000 nm, said method comprising:

- (a) dissolving the drug in a solvent;
- (b) adding the solubilized drug to a solution comprising at least one copolymer of vinyl pyrrolidone and vinyl acetate to form a clear solution; and
- (c) precipitating the solubilized drug having a copolymer of vinyl pyrrolidone and vinyl acetate as a surface stabilizer using a non-solvent,

wherein said method produces a nanoparticulate drug composition having at least one copolymer of vinyl pyrrolidone and vinyl acetate as a surface stabilizer and an effective average particle size of less than about 2000 nm.

5        29.      The method of claim 28, wherein the copolymer of vinyl pyrrolidone and vinyl acetate has from about 40% up to about 99.9% vinyl pyrrolidone and about 0.1% up to about 60% vinyl acetate.

10       30.     The method of claim 28, wherein the drug is present in an amount selected from the group consisting of from about 99.9% to about 0.1%, from about 80% to about 5.0%, and from about 50% to about 10%, by weight based on the total combined weight of the drug substance and surface stabilizer, not including other excipients.

15       31.     The method of claim 28, wherein the at least one copolymer of vinyl pyrrolidone and vinyl acetate is present in an amount selected from the group consisting of from about 0.1 to about 90%, from about 1 to about 75%, from about 10 to about 60%, and from about 10 to about 55%, by weight based on the total combined weight of the drug substance and surface stabilizer, not including other excipients.

20       32.     The method of claim 28, wherein the drug is selected from the group consisting of a crystalline phase drug and an amorphous phase drug.

25       33.     The method of claim 28, further comprising at least one surface stabilizer which is not a copolymer of vinyl pyrrolidone and vinyl acetate.

34.     The method of claim 28, comprising two or more copolymers of vinyl pyrrolidone and vinyl acetate.

30       35.     The method of claim 28, wherein the effective average particle size of the nanoparticulate composition is selected from the group consisting of less than about 1500

nm, less than about 1000 nm, less than about 800 nm, less than about 600 nm, less than about 400 nm, less than about 300 nm, less than about 200 nm, less than about 100 nm, and less than about 50 nm.

5        36.      A method of treating a patient in need with a nanoparticulate composition comprising an organic drug, wherein said method comprises:

(a)     administering to a mammal in need a therapeutically effective amount of the nanoparticulate composition having an effective average particle size of less than about 2000 nm and comprising:

10        (1)     an organic drug;

              (2)     at least one copolymer of vinyl pyrrolidone and vinyl acetate as a surface stabilizer adsorbed on the surface of the drug.

15        37.      The method of claim 36, wherein the copolymer of vinyl pyrrolidone and vinyl acetate has from about 40% up to about 99.9% vinyl pyrrolidone and about 0.1% up to about 60% vinyl acetate.

20        38.      The method of claim 36, wherein the drug is present in an amount selected from the group consisting of from about 99.9% to about 0.1%, from about 80% to about 5.0%, and from about 50% to about 10%, by weight based on the total combined weight of the drug substance and surface stabilizer, not including other excipients.

25        39.      The method of claim 36, wherein the at least one copolymer of vinyl pyrrolidone and vinyl acetate is present in an amount selected from the group consisting of from about 0.1 to about 90%, from about 1 to about 75%, from about 10 to about 60%, and from about 10 to about 55%, by weight based on the total combined weight of the drug substance and surface stabilizer, not including other excipients.

30        40.      The method of claim 36, wherein the drug is selected from the group consisting of a crystalline phase drug and an amorphous phase drug.

41. The method of claim 36, wherein the effective average particle size of the nanoparticulate composition is selected from the group consisting of less than about 1500 nm, less than about 1000 nm, less than about 800 nm, less than about 600 nm, less than  
5 about 400 nm, less than about 300 nm, less than about 200 nm, less than about 100 nm, and less than about 50 nm.